

New Opportunities for the Development of Education at the Technical University of Liberec

Specific objective A2: Development in the field of distance learning, online learning and blended learning

NPO_TUL_MSMT-16598/2022



KNT_TNA_Material conditions

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Funded by
the European Union
NextGenerationEU

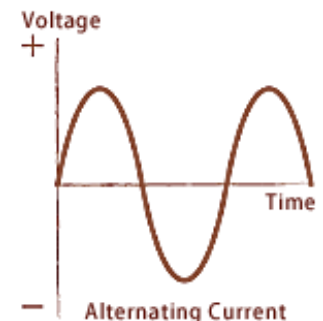
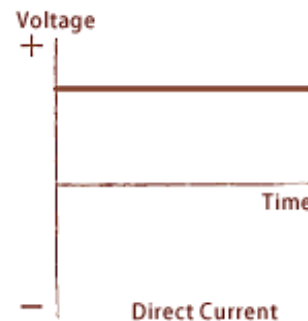


**CZECH
RECOVERY
PLAN**

MSMT
MINISTRY OF EDUCATION,
YOUTH AND SPORTS

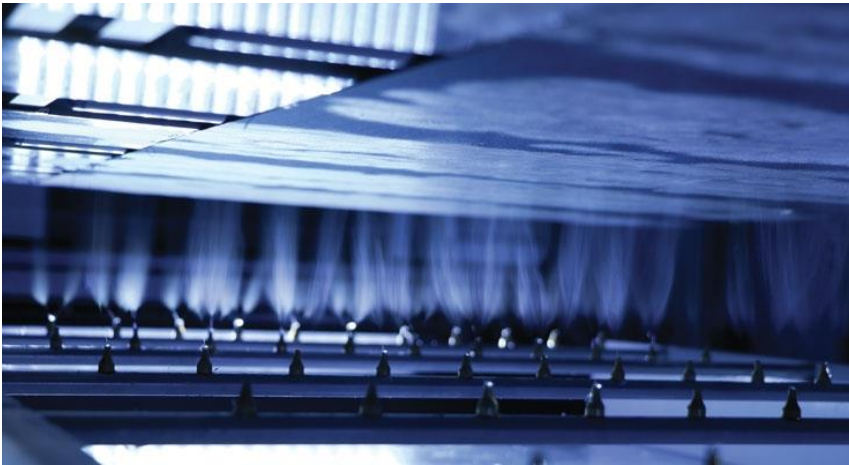
Repetition

- The process of fiber formation by the effect of an electric field
- Described as tugging between electrical and capillary forces
- High voltage DC power supply
 - It does not change its polarity over time
 - Positive / negative
- From solutions or melts

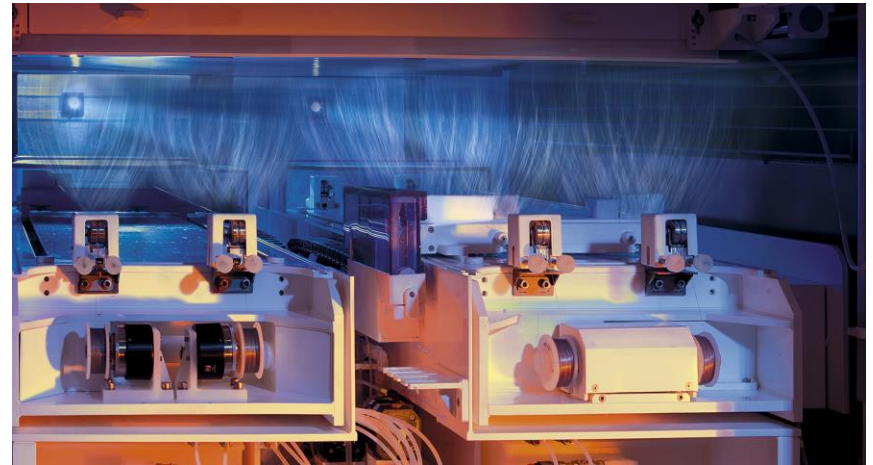


Repetition

Needle electrospinning



Needle-less electrospinning



Electrode - charged / grounded
Collector - charged / grounded

Conditions affecting electrospinning

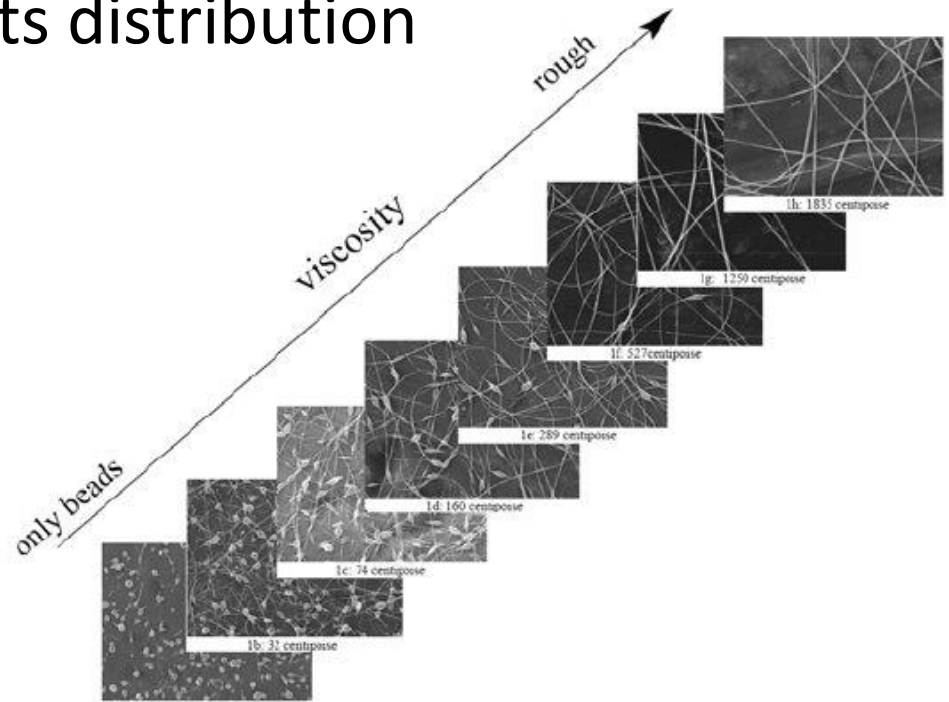
Process conditions \times material conditions

Process conditions

- Spinner configuration
- Voltage
- Distance of the collector from the electrode
- Dosage of the solution
- Background material
- Temperature
- Humidity

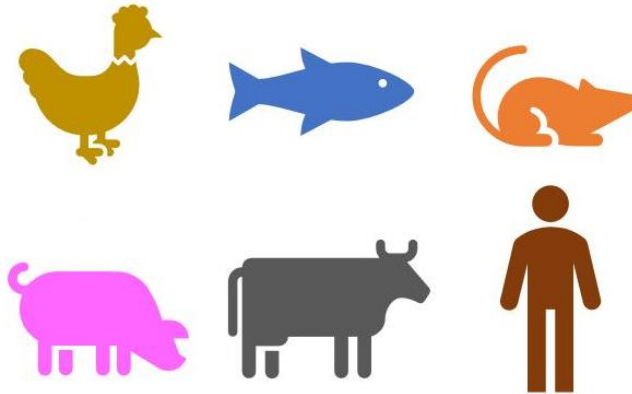
Material conditions

- Polymer type
- Molecular weight and its distribution
- Concentration
- Solvent system
- Electric conductivity
- Viscosity
- Surface tension
- Additives



Type of polymer

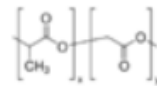
- Natural polymers
 - From what source?



- Synthetic polymers
 - Copolymers

Poly(D,L-lactide-co-glycolide)

7 Product Results | Match Criteria: Product Name



Synonym: **PLGA**

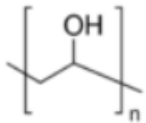
Linear Formula: $[\text{C}_3\text{H}_4\text{O}_2]_x[\text{C}_2\text{H}_2\text{O}_2]_y$

<input type="checkbox"/> P1941	lactide:glycolide (75:25), mol wt 66,000-107,000
<input type="checkbox"/> P2191	lactide:glycolide (50:50), mol wt 30,000-60,000
<input type="checkbox"/> P2066	lactide:glycolide 65:35, M_w 40,000-75,000

Type of polymer

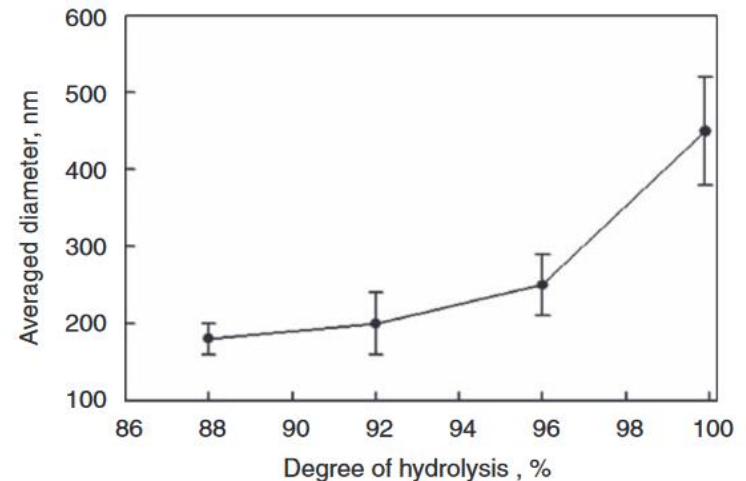
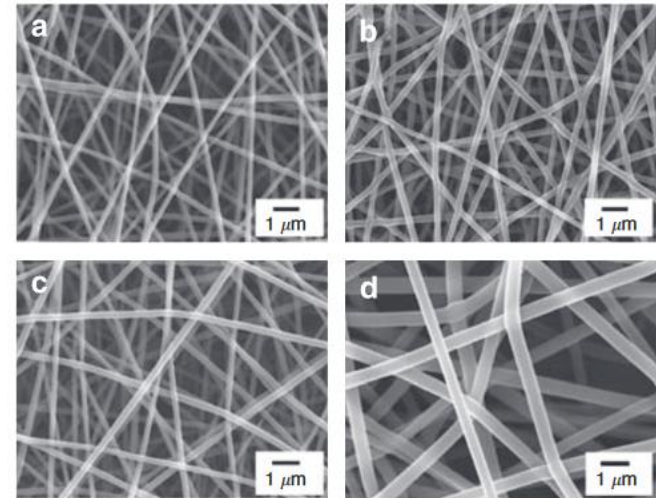
Poly(vinyl alcohol)

16 Product Results | Match Criteria: Product Name



Linear Formula: $[-\text{CH}_2\text{CHOH}-]_n$ | CAS Number: **9002-89-5**

<input type="checkbox"/> 341584	M_w 89,000-98,000, 99+% hydrolyzed
<input type="checkbox"/> P8136	87-90% hydrolyzed, average mol wt 30,000-70,000
<input type="checkbox"/> 360627	M_w 9,000-10,000, 80% hydrolyzed
<input type="checkbox"/> 363170	M_w 13,000-23,000, 87-89% hydrolyzed
<input type="checkbox"/> 363065	M_w 146,000-186,000, 99+% hydrolyzed
<input type="checkbox"/> 363146	M_w 85,000-124,000, 99+% hydrolyzed
<input type="checkbox"/> 363138	M_w 31,000-50,000, 98-99% hydrolyzed
<input type="checkbox"/> 363081	average M_w 85,000-124,000, 87-89% hydrolyzed



SEM images of electrospun PVA fibers with DH of (a) 88%, (b) 92%, (c) 96% and (d) 99.9% (top), and fiber diameters of the corresponding electrospun PVA fibers (bottom). The concentration of the PVA solution was 8.0wt%.

Molecular weight

- Mass of an amount of substance in a sample

$$M \text{ [g/mol]}$$

- Weight average molecular weight

$$M_w = \frac{\sum m_i M_i}{\sum m_i} = \frac{\sum n_i M_i^2}{\sum n_i M_i} = \sum w_i M_i$$

- Numerically average molecular weight

$$M_n = \frac{\sum m_i}{\sum n_i} = \frac{\sum n_i M_i}{\sum n_i} = \sum x_i M_i$$

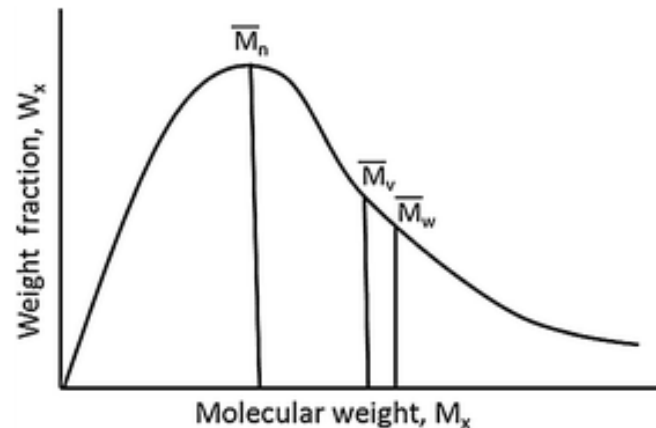
Molecular weight

- A mixture of molecular weight samples:

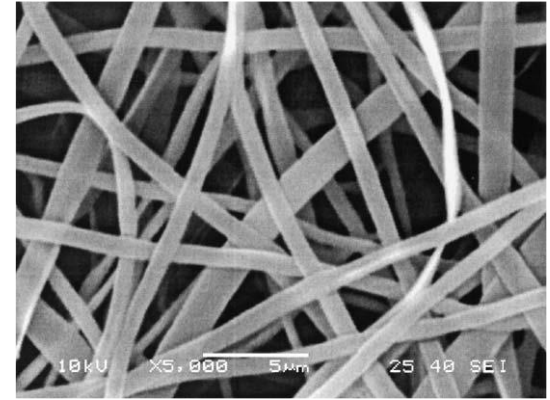
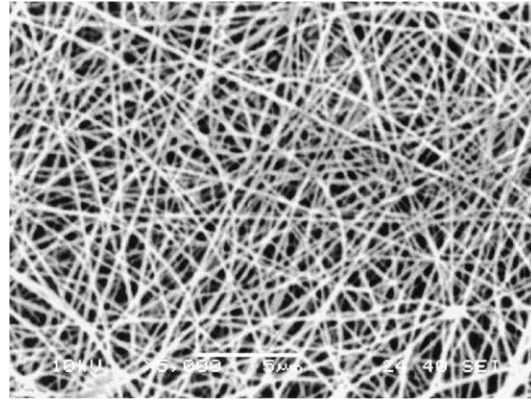
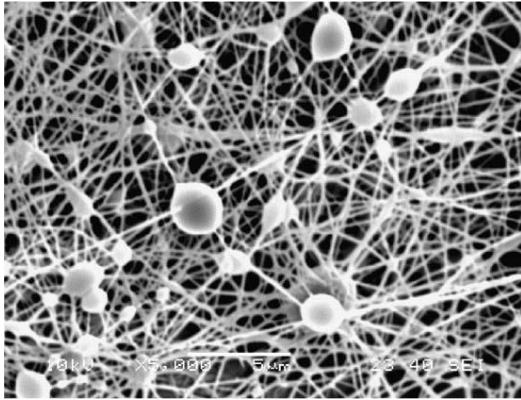
10 000; 10 000; 10 000; 100 000

- M_n 32 500; M_w 79 230

- Molecular weight distribution



Molecular weight

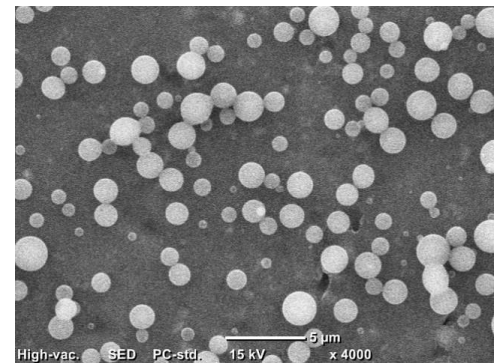
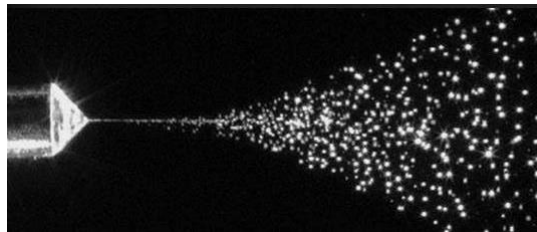
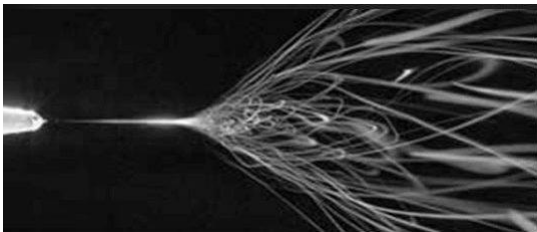


PVA (A) 9000–10,000 g/mol; (B) 13,000–23,000 g/mol; and (C) 31,000–50,000 g/mol (solution concentration: 25 wt.%).

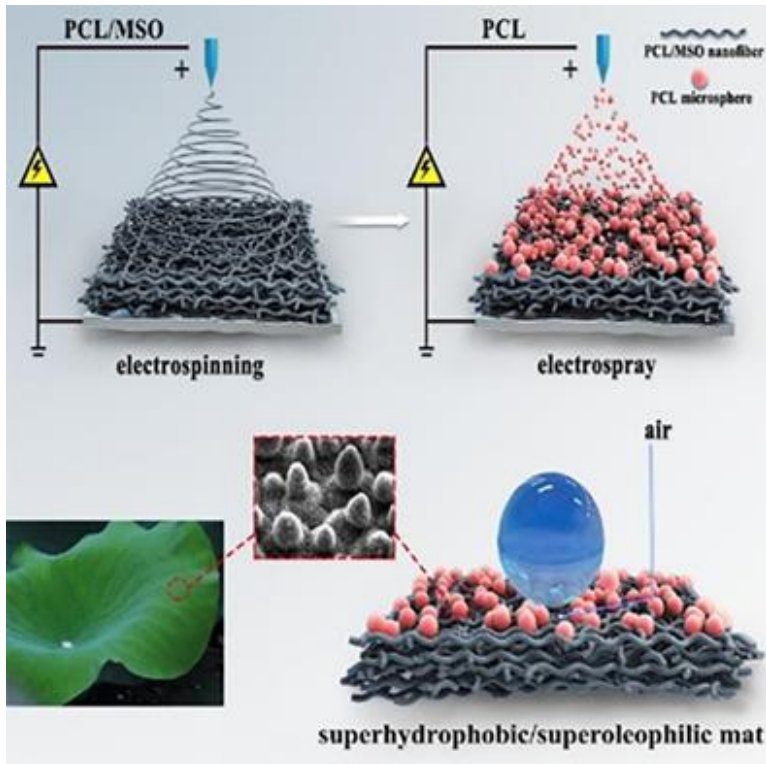
- It affects fiber diameters
- It is necessary to adjust the concentration of the solution

Concentration

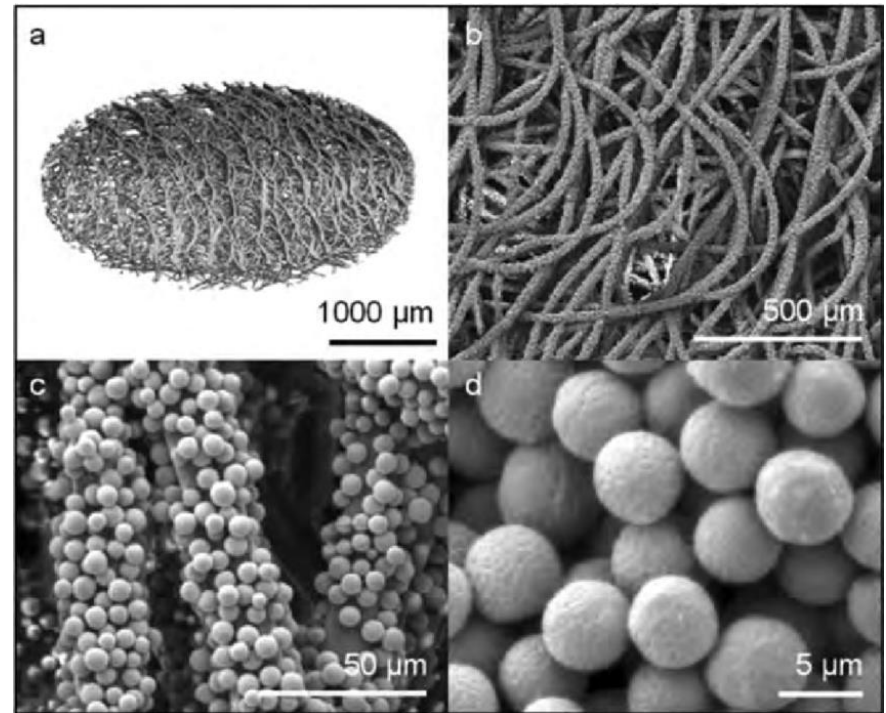
- An important property that affects the electrospinning process
- It affects viscosity, surface tension and electrical conductivity
- With a low concentration, we can create beads
- Electrospaying



Concentration



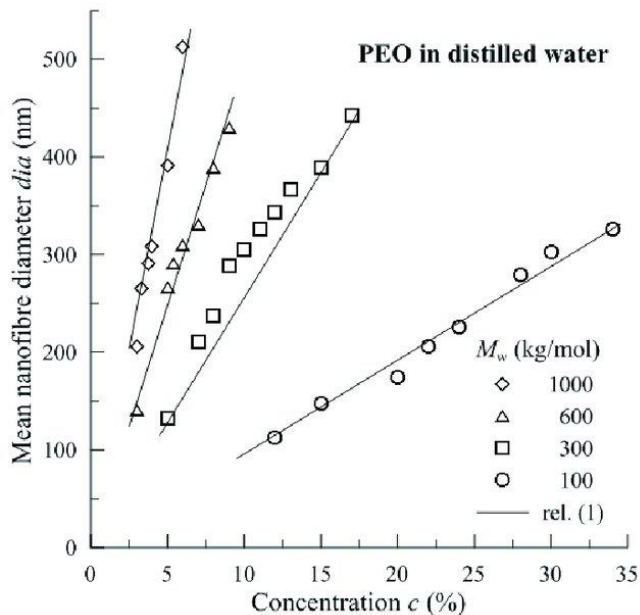
Schematic mechanism of superhydrophobic surface in PCL/MSO-PCL composite membrane



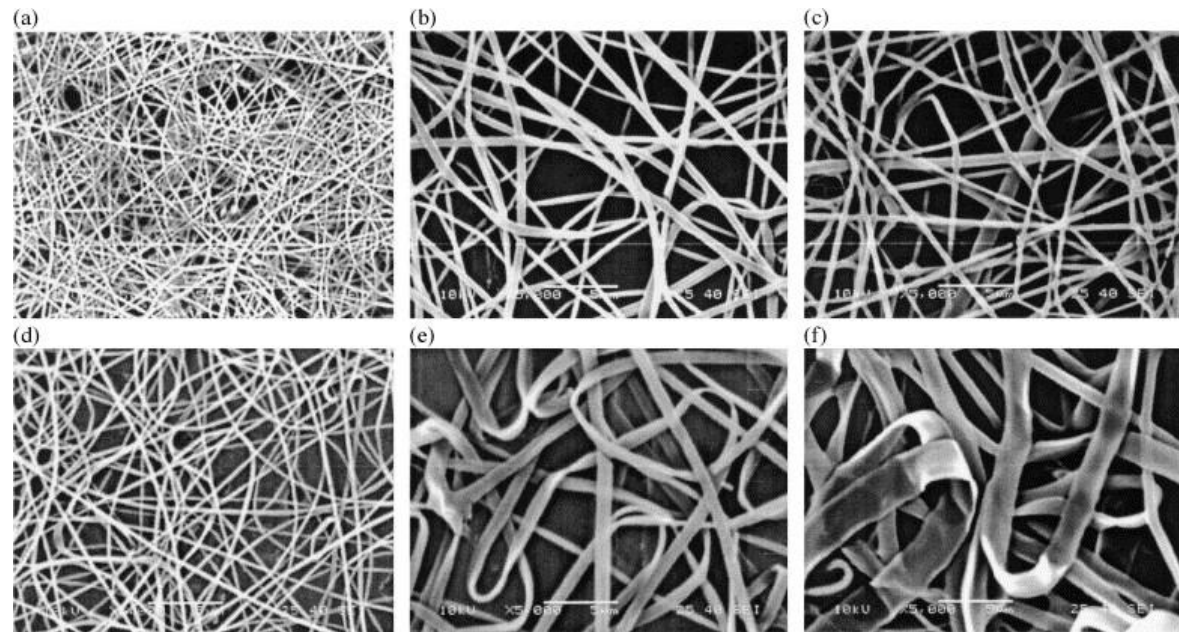
Overview of a PCL scaffold coated with electrospayed PLGA particles after 1 h of electrospaying (80 mg PLGA). (a) mCT, (b)-(d) SEM, at different magnifications.

Concentration

- By changing the concentration, the fiber diameters change

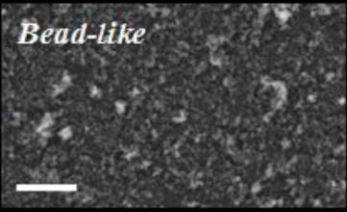

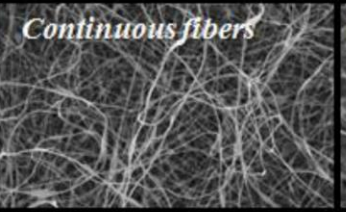
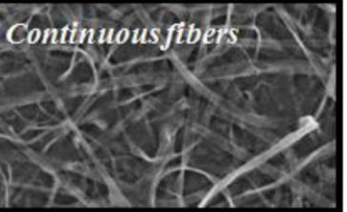
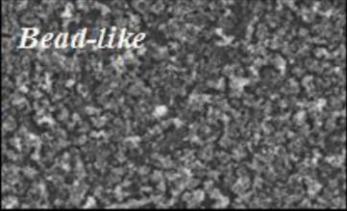


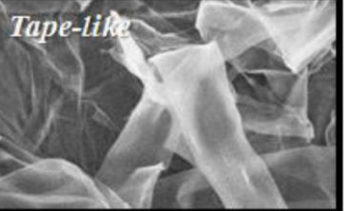
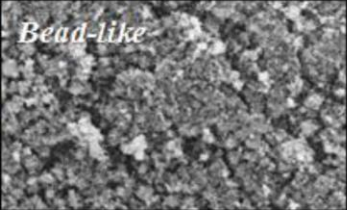
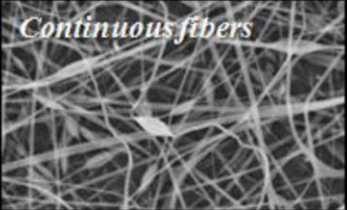
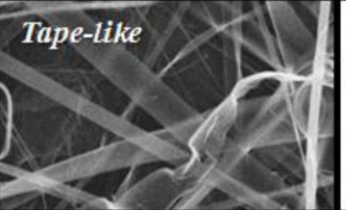
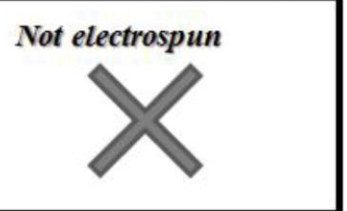
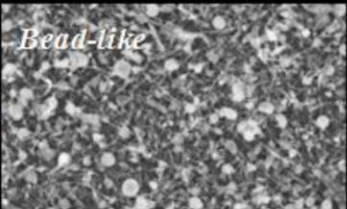
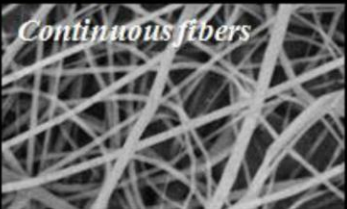
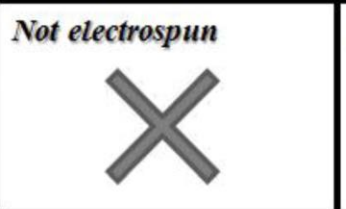
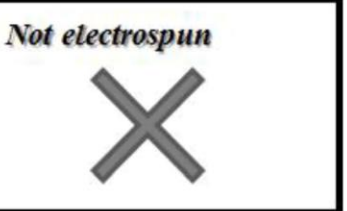


Dependence of mean nanofiber diameter on concentration (c) and molecular weight (M_w).



Photographs showing the effect of solution concentration on the structure in the electrospun polymer. Molecular weight=13,000–23,000 g/mol; (a) 21 wt.%; (b) 27 wt.%; and (c) 31 wt.%. Molecular weight=50,000–89,000 g/mol; (d) 9 wt.%; (e) 13 wt.%; (f) 17 wt.%.

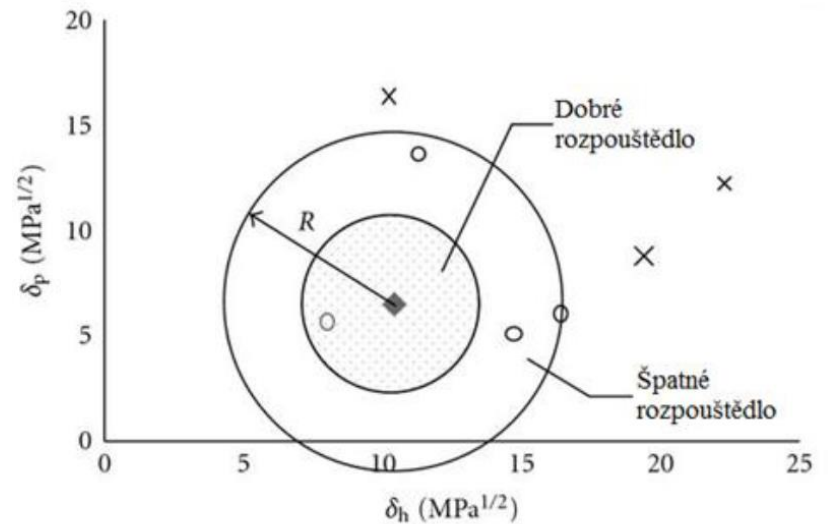
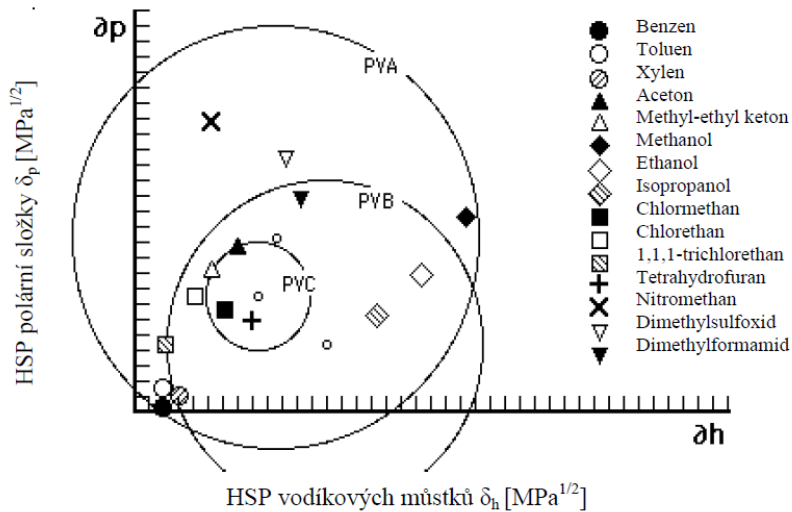
Concentration

		Molecular weight (M_n)			
		10,200	17,100	231,000	538,800
Polymer concentration (wt%)	1.0	<i>Bead-like</i> 	<i>Bead-like</i> 	<i>Continuous fibers</i> 	<i>Continuous fibers</i> 
	3.0	<i>Bead-like</i> 	<i>Fibers with beads</i> 	<i>Continuous fibers</i> 	<i>Tape-like</i> 
	5.0	<i>Bead-like</i> 	<i>Continuous fibers</i> 	<i>Tape-like</i> 	<i>Not electrospun</i> 
	10.0	<i>Bead-like</i> 	<i>Continuous fibers</i> 	<i>Not electrospun</i> 	<i>Not electrospun</i> 

Effects of molecular weight and polymer concentration on the PNIPAAm fiber formation and morphologies (scale bar 10 μ m).

Solvent system

- Polymer and solvent solubility parameters
- Hansen solubility parameters



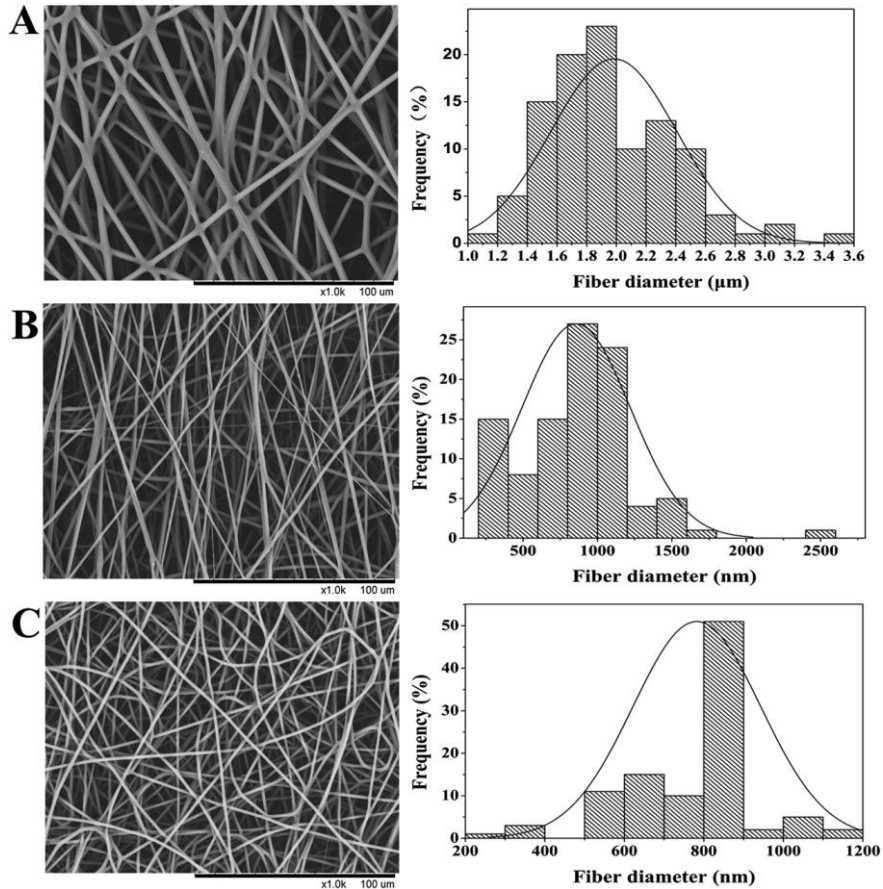
- ◆ Polymer
- Rozpouštědlo
- × Nerozpouštědlo

Solvent system

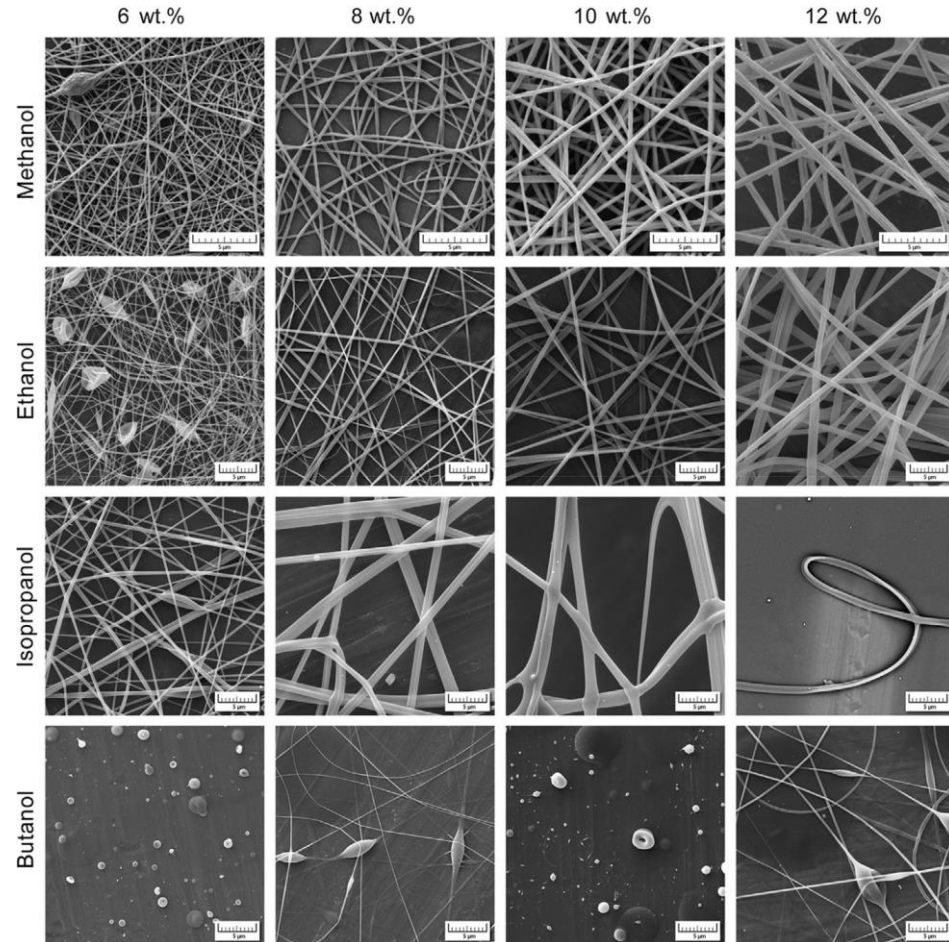
- Solubility parameters of polymer and solvent
- It affects electrical conductivity, viscosity and surface tension
- It therefore affects the electrospinning process and the morphology of the fiber layer
- Choice with respect to subsequent application



Solvent system



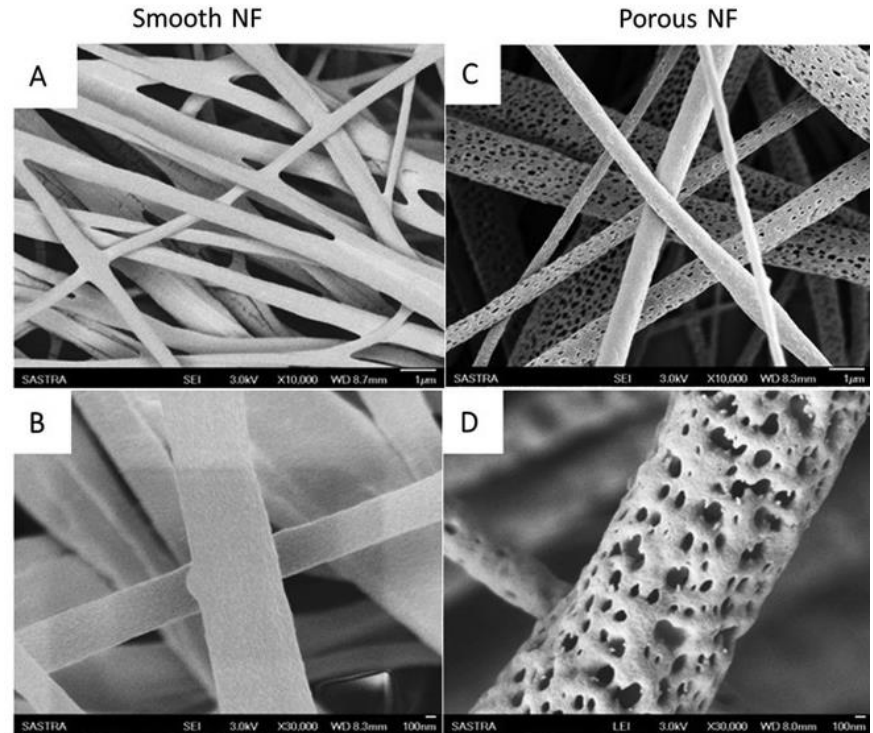
SEM images of PHBV/PEO electrospun fibers with fiber diameter distribution obtained from 10% (w/v) PHBV/PEO solution prepared by using different solvents: (A) Chloroform (B) Dichloromethane (C) Dichloromethane/ethanol.



SEM images of PVB nanofibers

Solvent system

- Influencing surface morphology
- Possibility of creating porous nanofibers
- Very fast evaporation of the solvent
- Chloroform, acetone,...



PLA nanofibers. The co-solvent mixture DCM/DMF yielded smooth structures (Fig. 3a, b) while DCM/acetone solvent mixture led to the formation of heterogeneous porous nano-features on nanofibers (Fig. 3c, d).

Viscosity

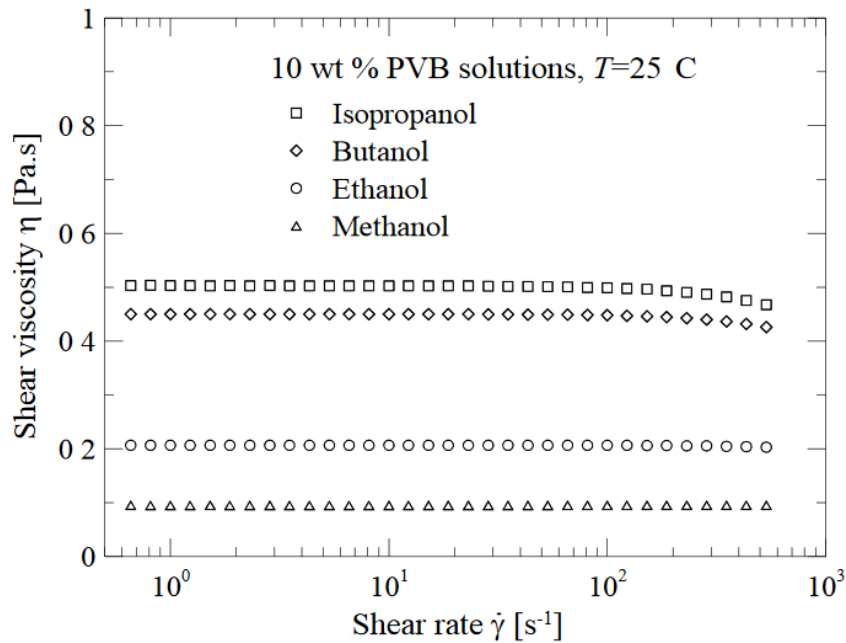
- Characterizes the internal friction of the fluid η [Pa.s]

- Dynamic viscosity $\tau = \eta \frac{dv}{dy}$

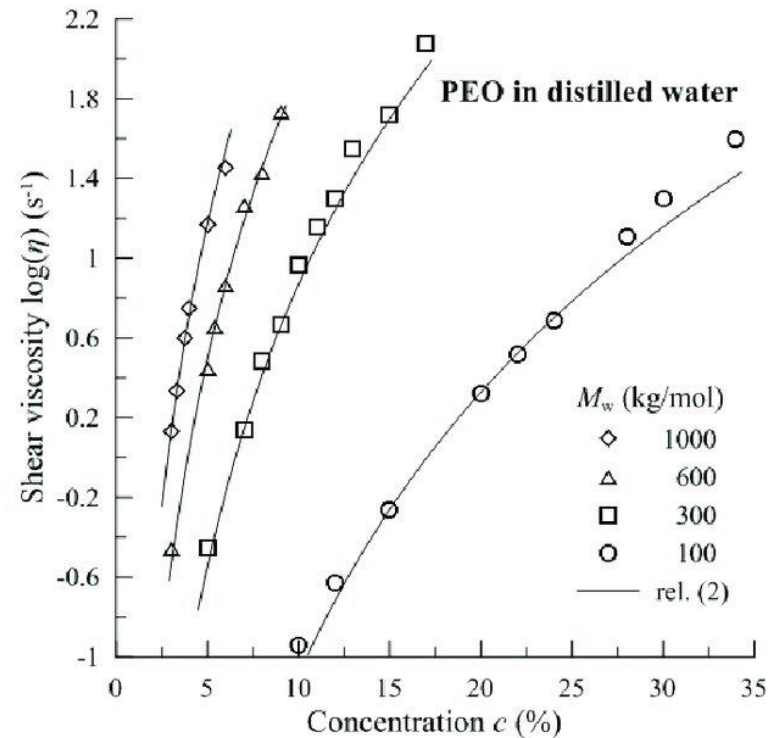
- Kinematic viscosity $\nu = \frac{\eta}{\rho}$

- It affects:
 - Molecular weight
 - Concentration
 - Solvent system

Viscosity

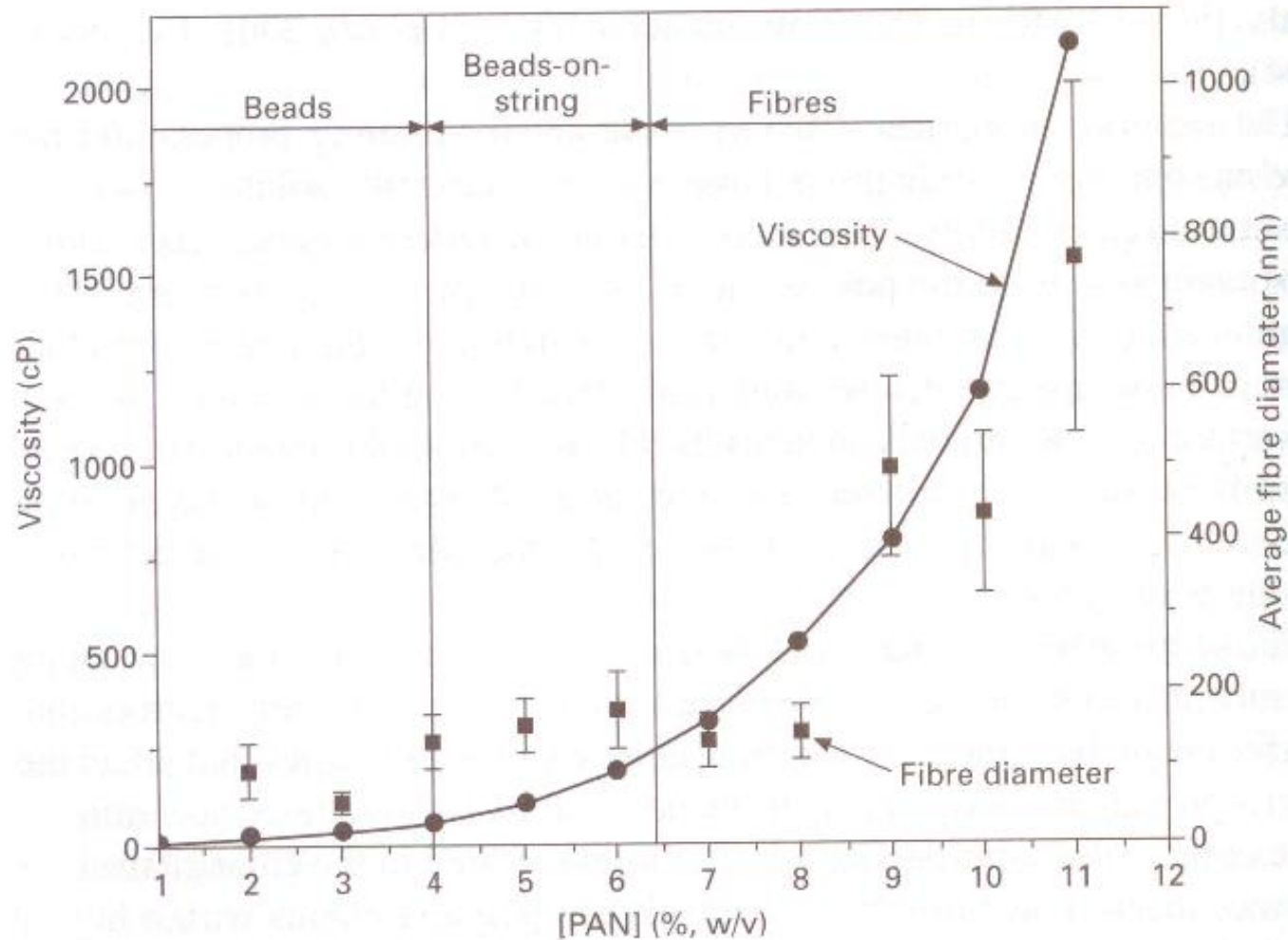


Shear viscosity of PVB solutions



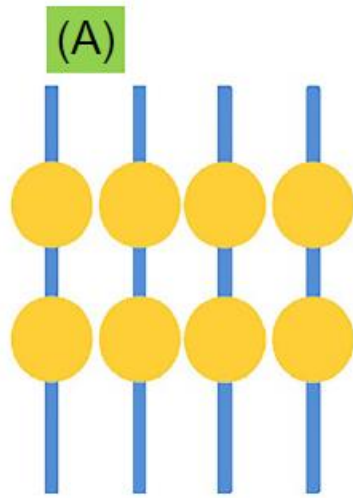
Dependence of shear viscosity ($\log(\eta)$) on concentration (c) and molecular weight (M_w) in semi-log coordinates.

Viscosity

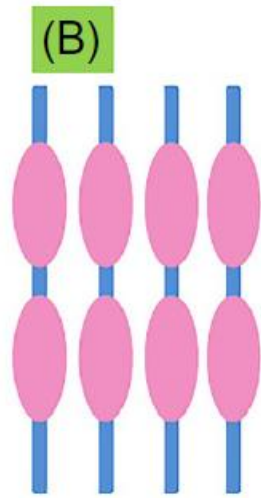


Relationship between polymer concentration, solution viscosity and diameter of electrospun polyacrylonitrile (PAN) fibres.

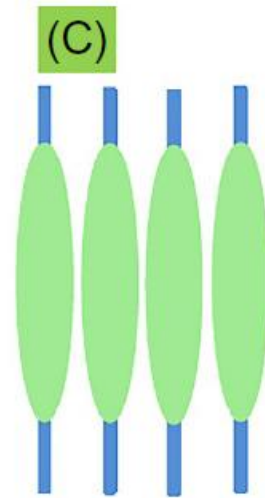
Viscosity



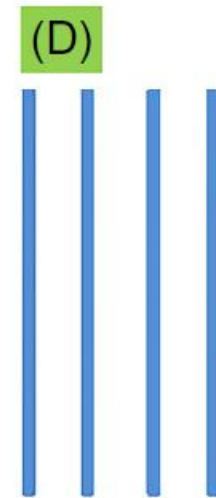
Droplet shape



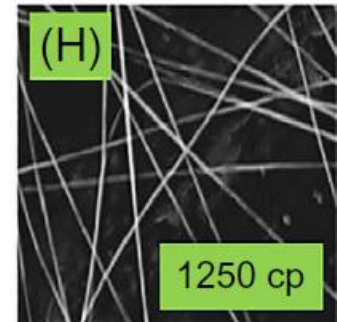
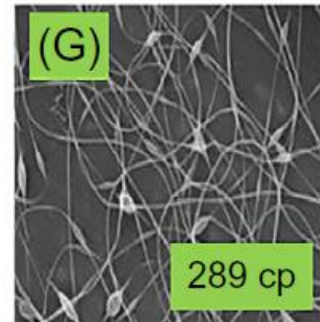
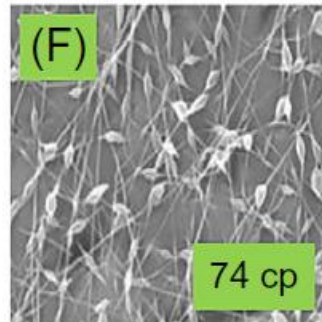
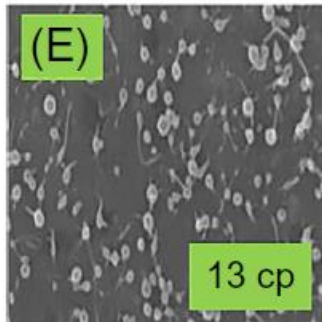
Elongated droplet



Stretched droplet



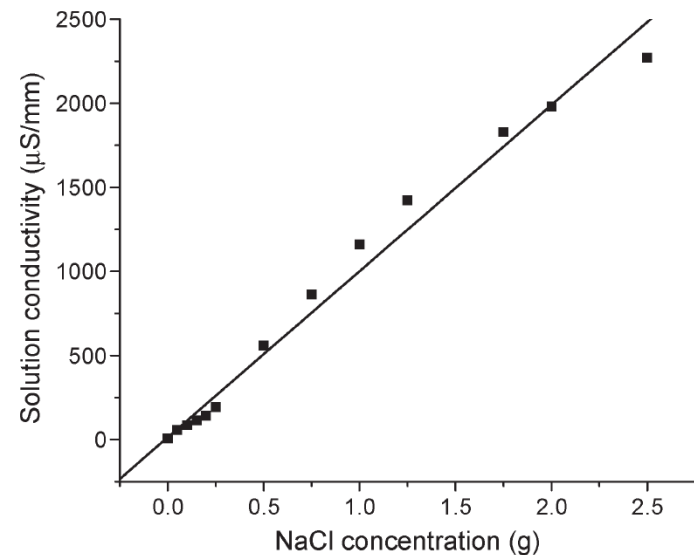
Nanofibers



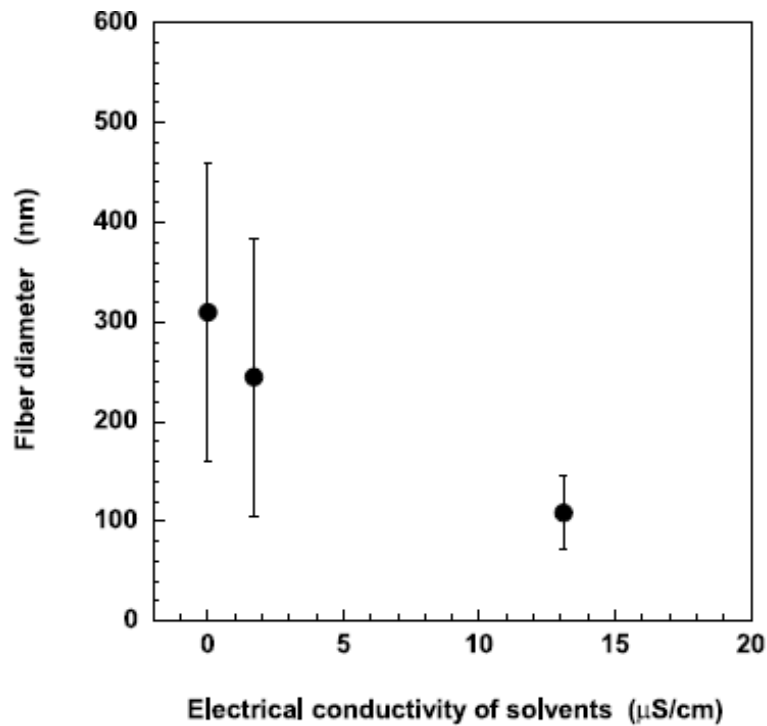
Electric conductivity

- The ability of a substance to conduct electric current σ [S/m]
- It affects:
 - Solvent system
 - Concentration
 - Additives

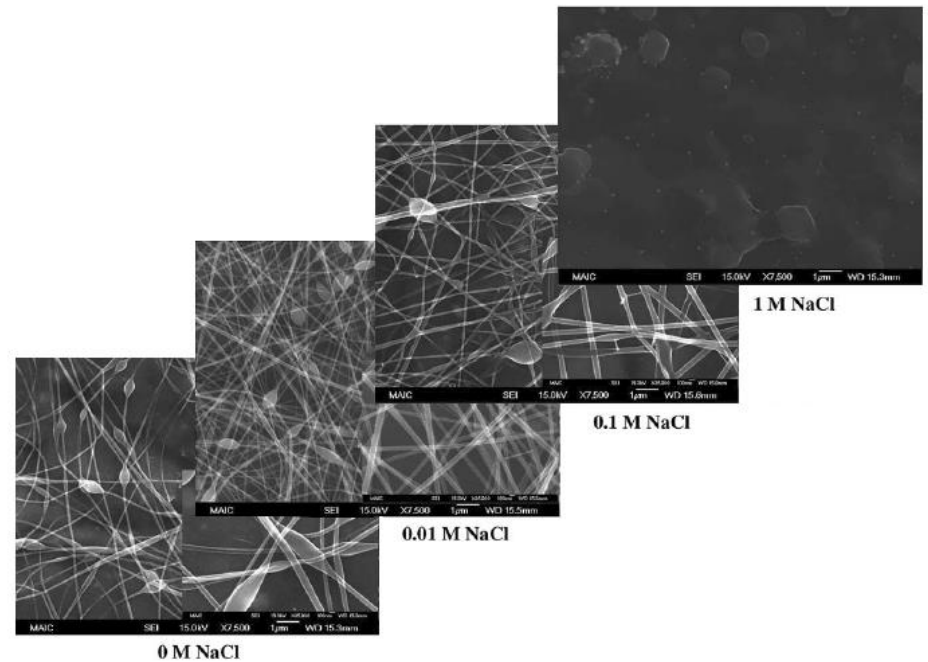
	Relative permittivity [-]	Specific conductivity [S/m]	Surface tension [mN/m]	Density [g/cm^3]
Methanol	32.7	$1.5 \cdot 10^{-7}$	22.12	0.7899
Ethanol	24.5	$1.35 \cdot 10^{-7}$	21.9	0.785
Isopropanol	19.9	$58 \cdot 10^{-7}$	21.38	0.7813
Butanol	17.5	$9.12 \cdot 10^{-7}$	24.5	0.806
PVB	3.60	$1 \cdot 10^{-9}$	-	1.09



Electric conductivity



Solution conductivity effects on the diameter of electrospun P(LLA-CL) (70/30 wt%) fibers



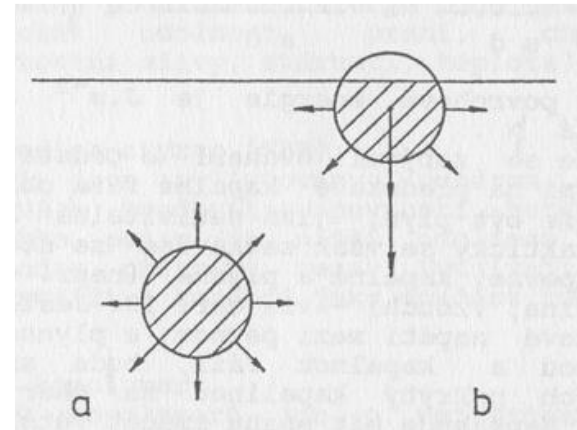
SEM images of PAA nanofibers with different NaCl concentrations

Surface tension

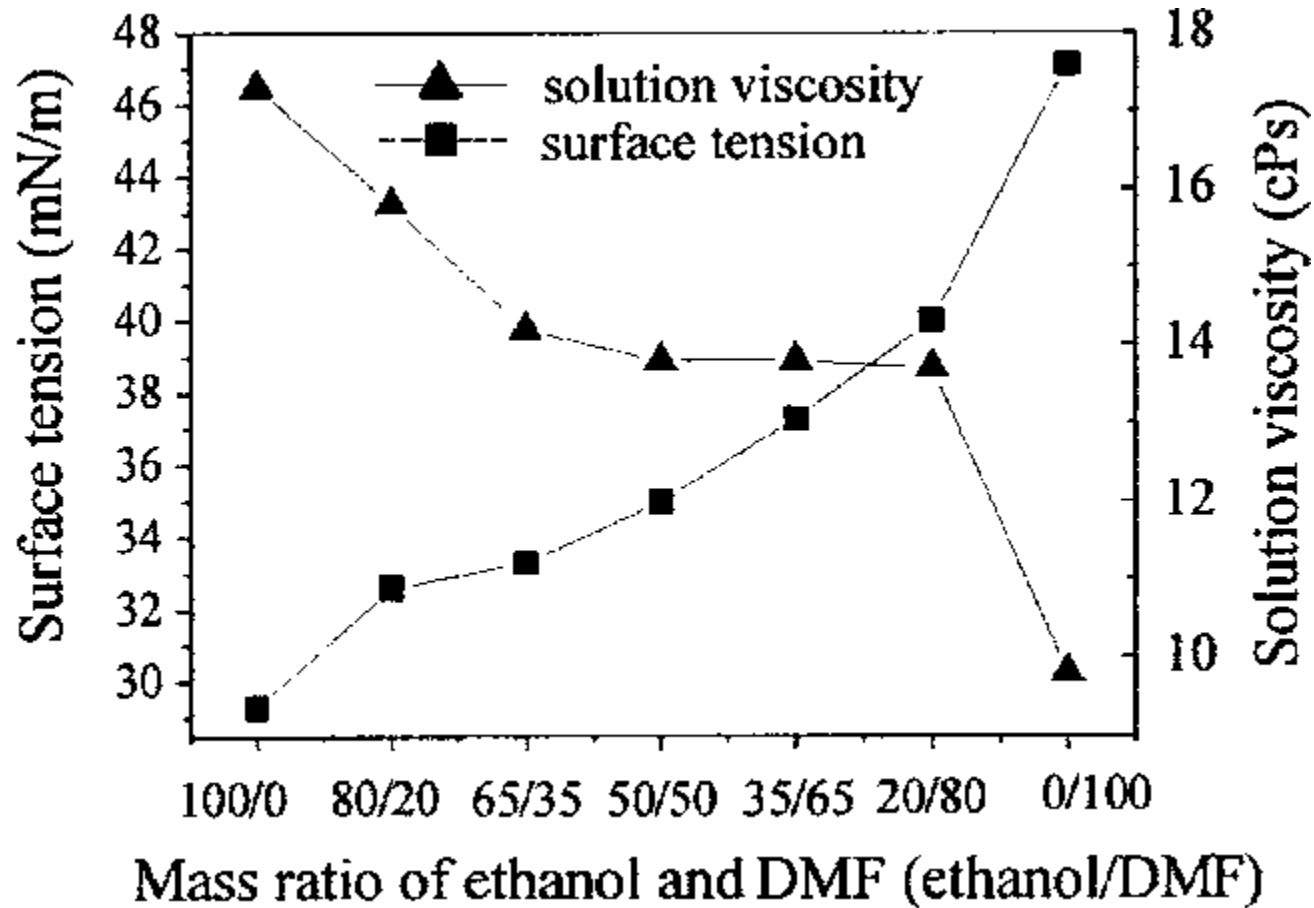
- The consequence of the interaction of the forces of the molecules or atoms that make up the surface layer
- The liquid tries to occupy the smallest possible area

- Affected by:
 - Solvent system
 - Concentration
 - Additives

$$\sigma = \frac{\Delta F}{\Delta l}$$

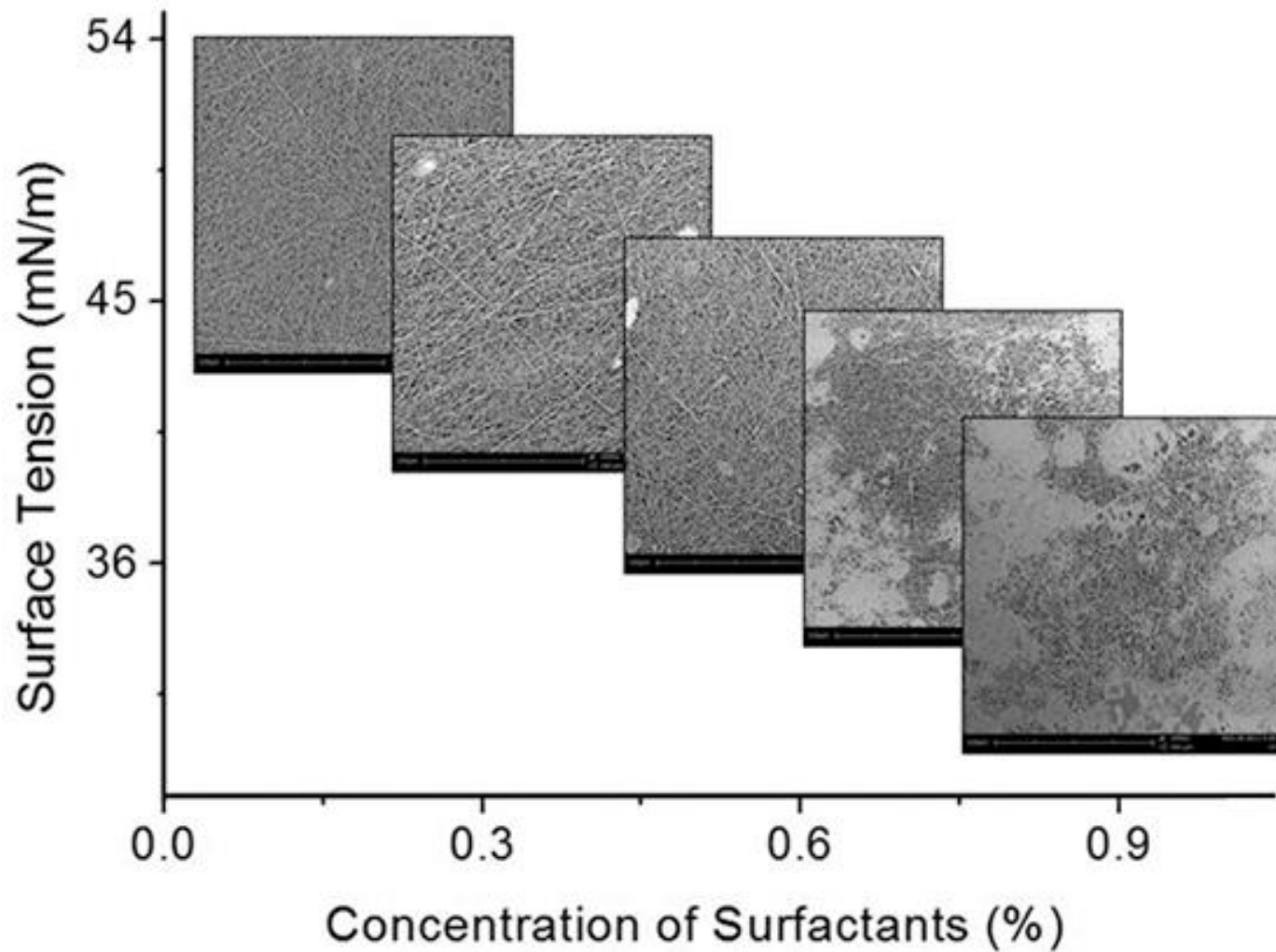


Surface tension



Surface tension and viscosity as functions of mass ratio of ethanol/DMF.

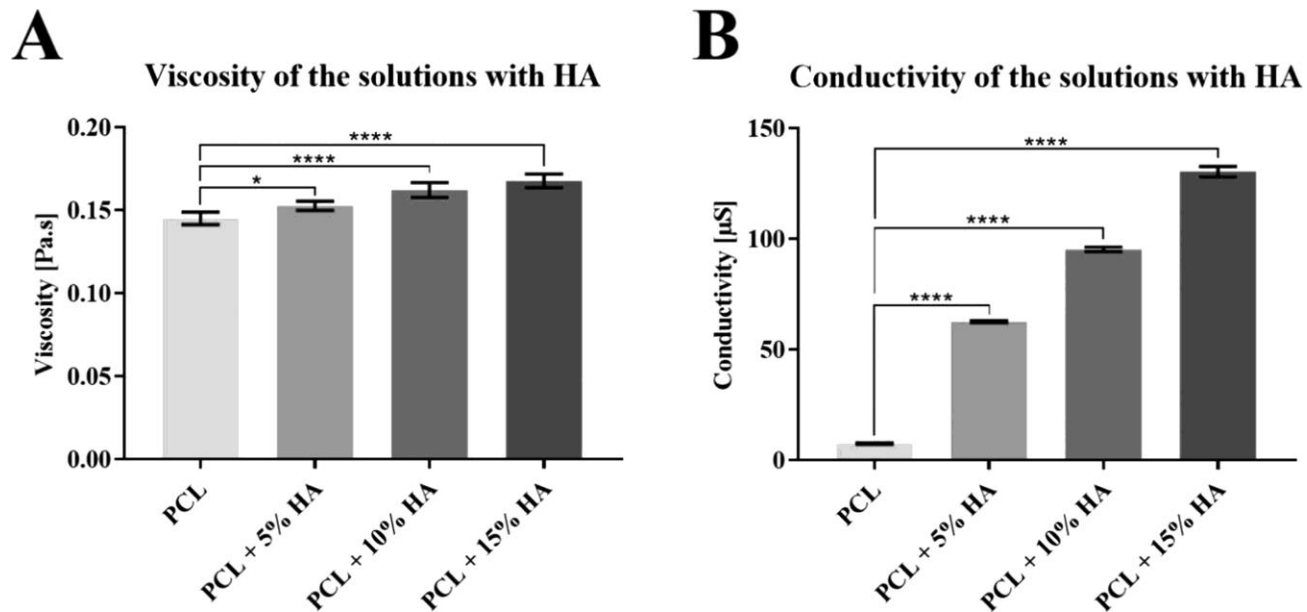
Surface tension



Effect of surface tension on the surface morphology of nanofibers

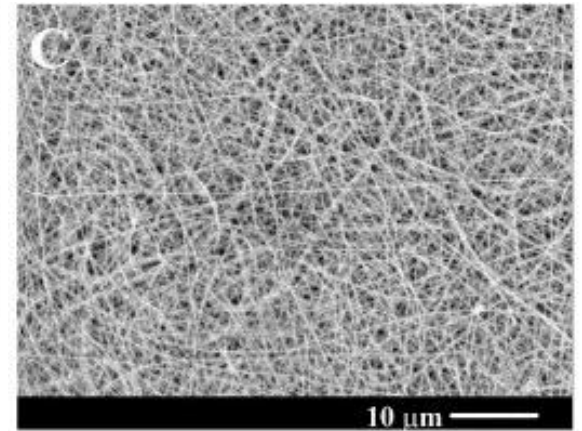
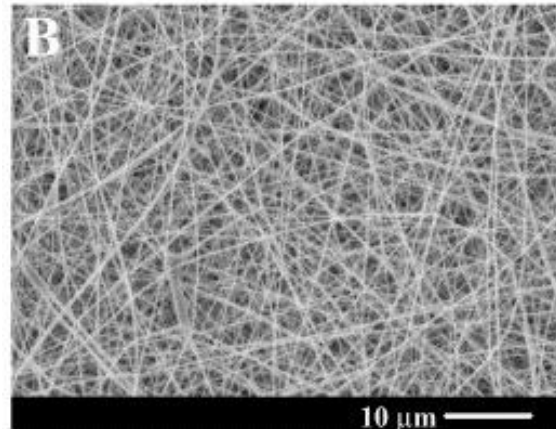
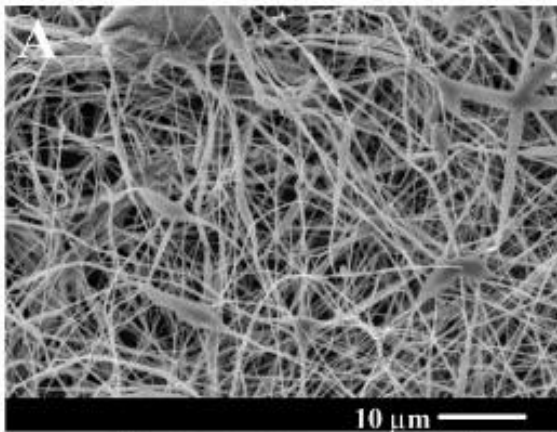
Additives

- They affect the surface tension, viscosity, electrical conductivity of the polymer solution
- All substances that are added to the solution



Parameters of the prepared PCL solutions with HA: (A) viscosity and (B) conductivity of the solutions.

Additives

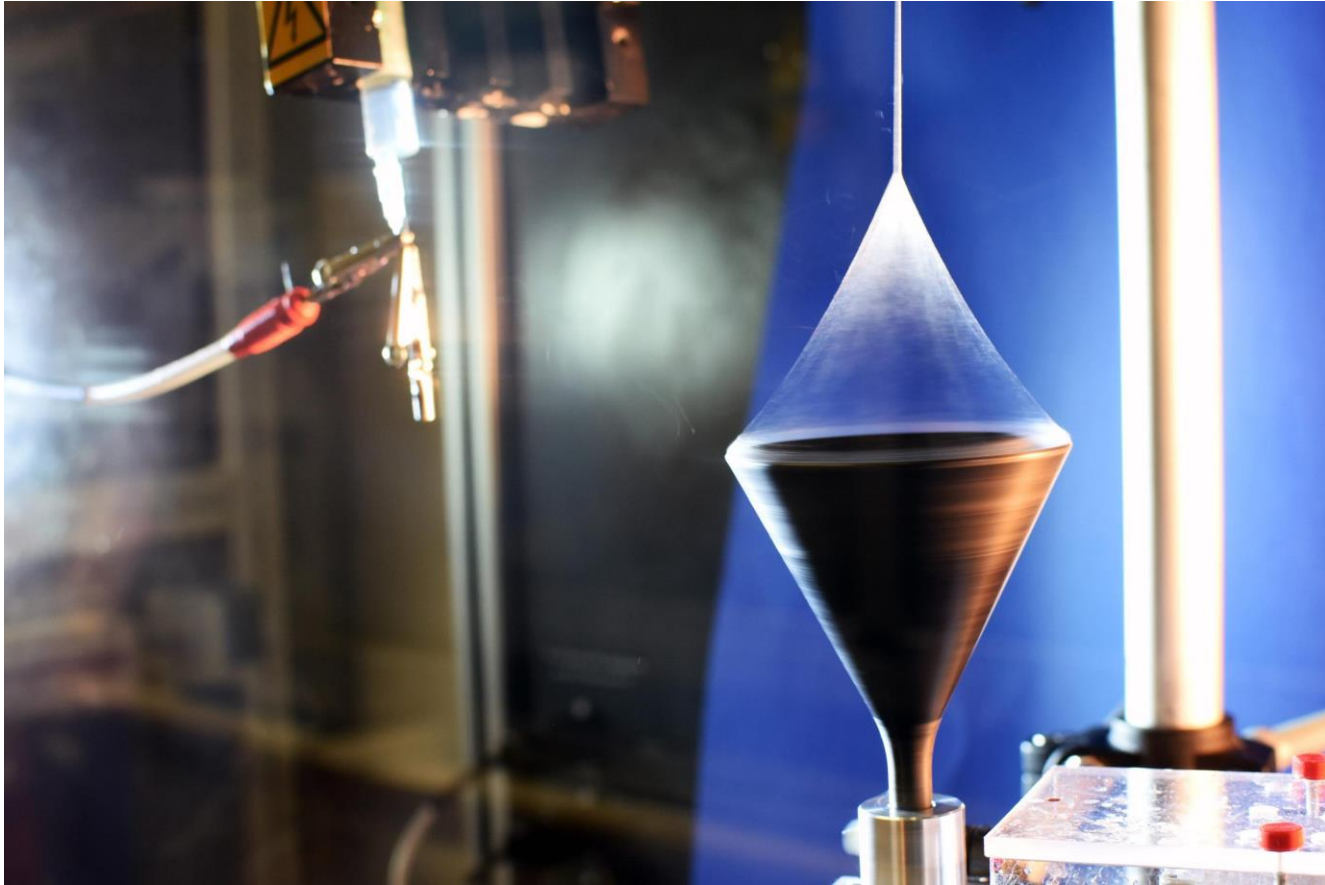


SEM images of PDLA membranes fabricated by electrospinning of a 30 wt% solution at voltage of 20 kV, feeding rate of 20 $\mu\text{l}/\text{min}$ and with 1 wt% of (A) KH_2PO_4 ; (B) NaH_2PO_4 and (C) NaCl .

Summary

- The properties of the solution affect the electrospinning process
- For optimal spinning it is necessary:
 - Find a match between the molecular weight and the concentration of the polymer in the solution
 - Choose a suitable solvent system
 - Ensure optimal conductivity and surface tension for the given polymer and solvent system

Thank you for your attention!



TEST

- What material conditions affect the spinning process?
- How does molecular weight affect the concentration of a polymer solution?
- How does the concentration of the polymer solution affect the diameter of the fibers?