Textilní chemie



13. Směsi vláken



Jakub Wiener



Jeden druh vlákna = jedny užitné vlastnosti ... jedna cena ... jedna ideální aplikace

polyester – ideální pro aplikace vyžadující vysokou pevnost, chemickou odolnost, dlouhou životnost, malou navlhavost, nemačkavost

bavlna – ideální pro vysokou navlhavost, smáčivost, bobtnavost, ekologičnost, barvitelnost, mačkavost, malá odolnost v oděru





Vlákna se využívají velmi často v podobě směsí

Pro mechanickou technologii je to snadné – na chemickém složení vláken nezáleží. Jde to snadno, tak proč to nedělat. Bude to levnější. Bude to šikovnější pro některé aplikace.



Směsování vláken



1) Při chemické úpravě (zejména barvení) textilií

2) Při deklaraci složení textilie

- 3) Při údržbě textilií
- 4) Při recyklaci
- 5) Při analýze textilií

Směsi vláken a komfort



směsování



Běžné směsi pro oděvní aplikace Polyester/bavlna, polyester/akryl, polyester/vlna a polyester/viskóza:

nemačkavý a odolný polyester se přidává k mačkavým přírodním vláknům (např. bavlně), výsledná tkanina je odolnější.

Praní: šetrný prací program, ruční praní nebo praní v pračce při 40 °C. Dopředu nenamáčejte. Neždímejte.

Žehlení: Žehlete opatrně.



Směsi vláken a hoření



Co z toho dokážete zapálit?



parafin



kevlar



svíčka









Kevlar



svíčka



Směsi vláken a hoření



Proč svíčka hoří???



Melanže



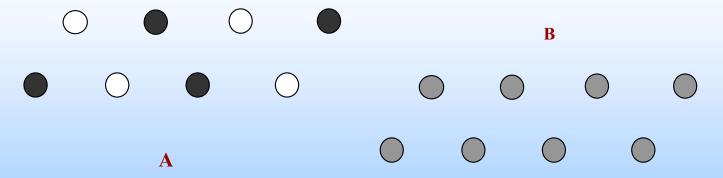


Směsování různobarevných vláken

Zajímavý vzhled



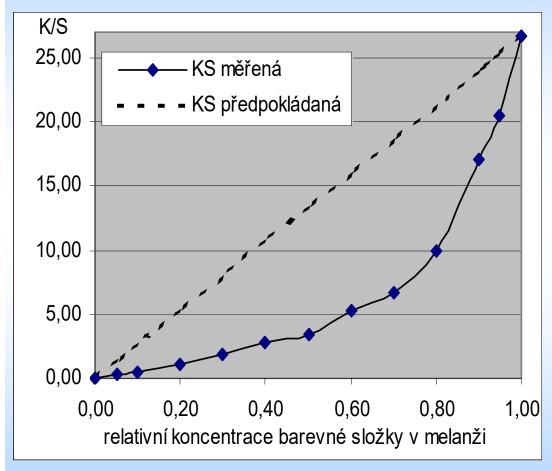
Pokud smícháme v poměru 1:1 barevná a nebarevná vlákna, bude výsledná směs jako celek obsahovat 50% barviva vůči textilii pouze z barevných vláken. Výsledná melanžová textilie bude výrazně světlejší, než by byla textilie obsahující stejné množství barviva rovnoměrně rozděleného mezi všechna vlákna





Vliv melanžování na sílu odstínu





K/S-hodnoty melanžové textilie jako funkce jejího složení - jedná se o směs černé a nebarvené viskózové stříže v různých poměrech – ve srovnání se stejnoměrně obarvenou viskózovou stříží na šedé až černé odstiny (čárkovaně – poskytuje lineární průběh).

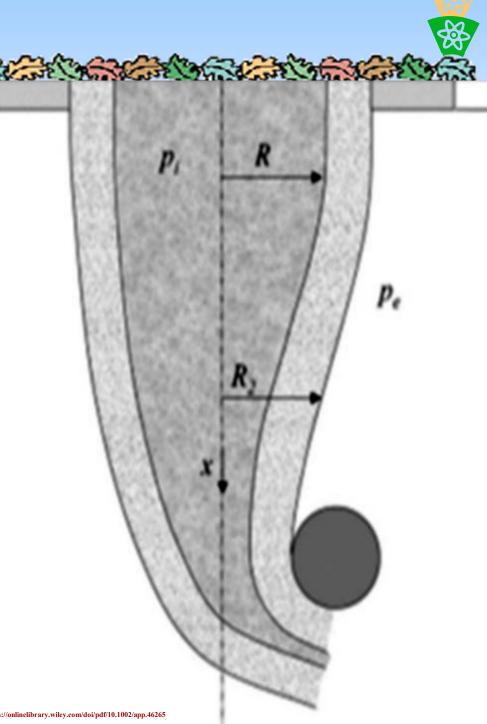
Bikomponentní vlákna

= směsování na úrovni polymerů

Jen pro tavitelné polymery

Princip: jednou zvlákňovací tryskou prochází současně několik polymerů tak, aby nedošlo k jejich smísení.

Lze použít polymerů s odlišnou teplotou tání, s rozdílnou polaritou, rozpustností ...



Užití bikomponentních vláken

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Speciální průřezy vláken

Extrémně jemná vlákna

Pojivá vlákna (netkané textilie)

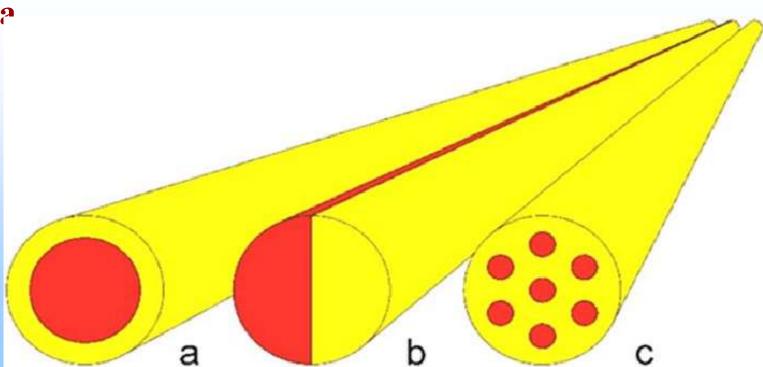
Barvitelnost vláken (PES v PP)

Optická vlákna

Biokompaktibilit?

Obloučkovitost

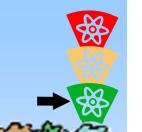
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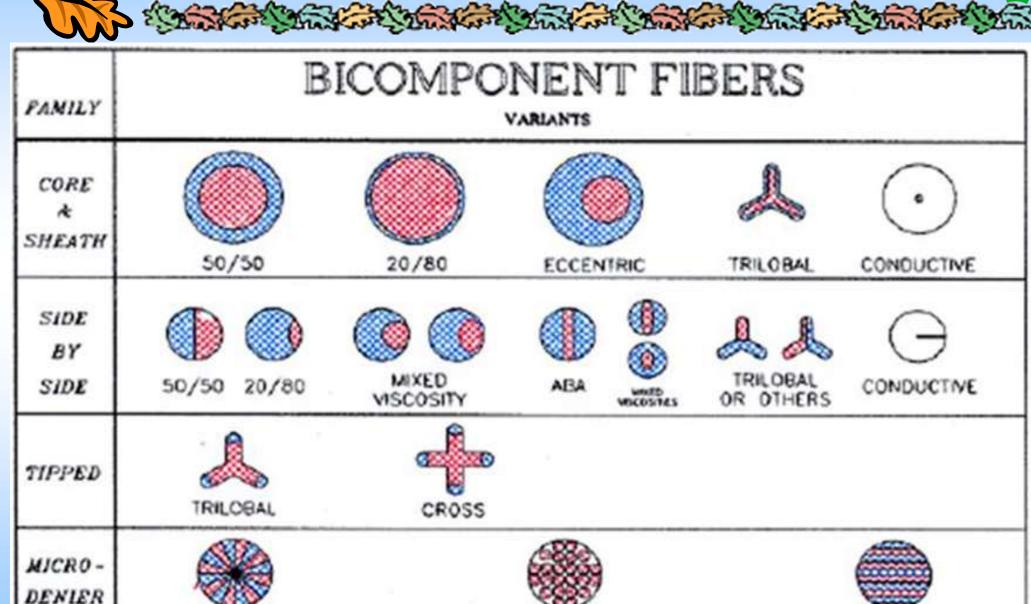


SEGMENTED PIE

Druhy bikomponentních vláken



STRIPED



ISLANDS-IN-A-SEA



Výroba vláken

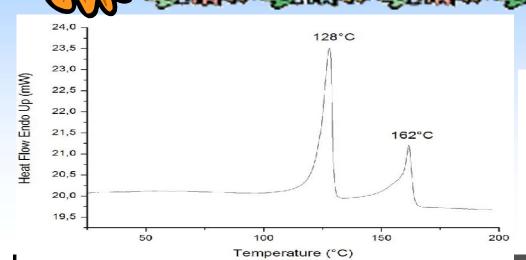






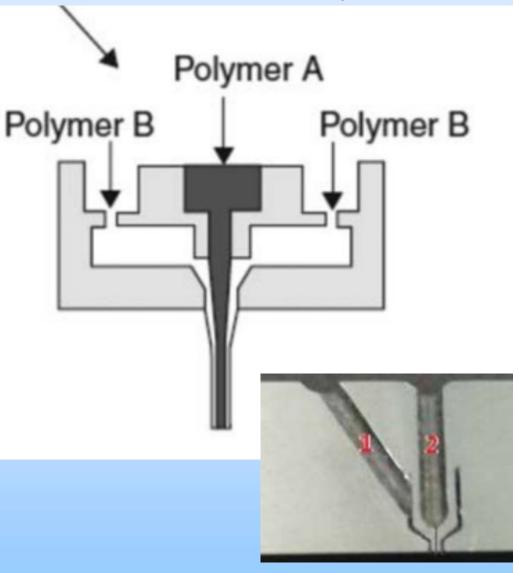
PE/PP bikomponentní vlákna

pro pojení

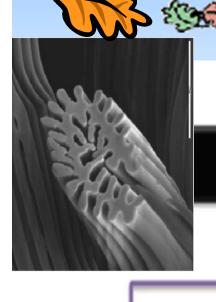


Kde je PP? Kde je PE?

Zvlákňovací hlava s trykou

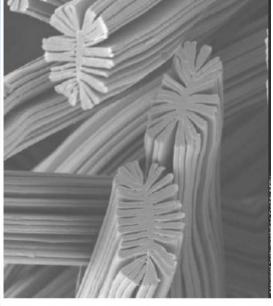


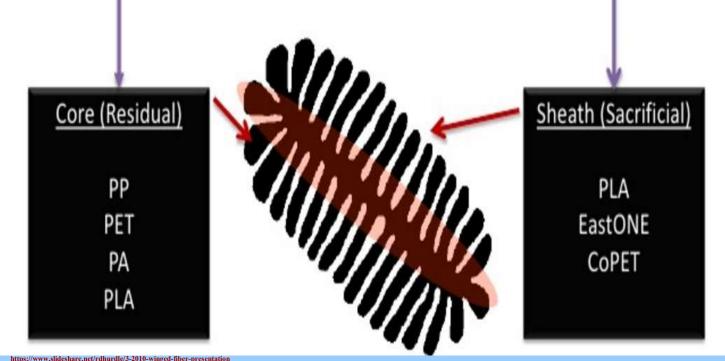
Bikomponentní vlákna pro filtrace

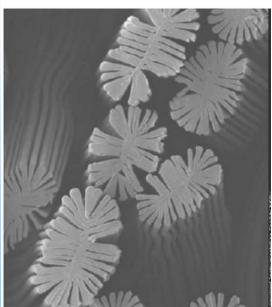


Sheath-Core Configuration







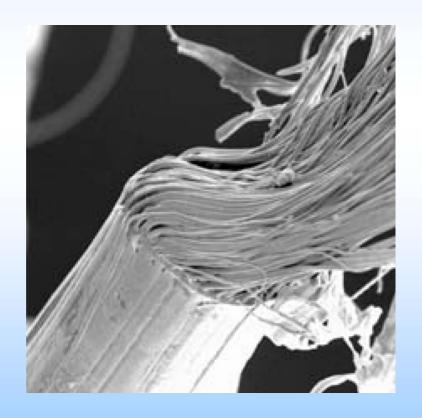


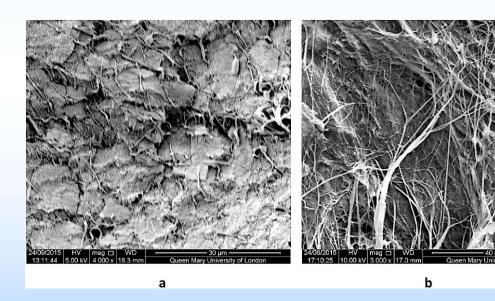


"Ostrovy v moři" bikomponentní vlákna



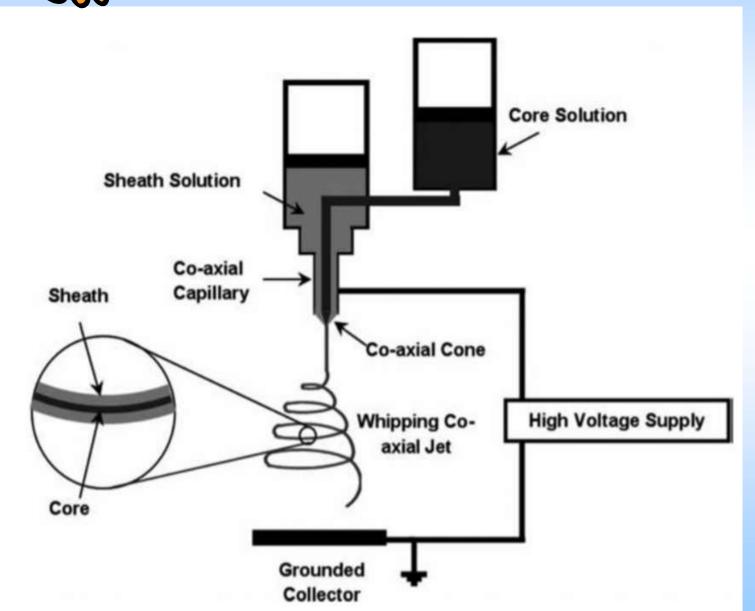




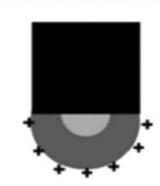


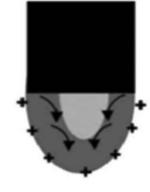


Bikomponentní nanovlákna















Analysis of fibre mixtures



Toxicity of chemicals used



Acetone - flammable, irritant, skin drying, LD50, oral, rat(mg.kg-1): 5800 mg/kg Chlorate - corrosive, oxidising, highly toxic to aquatic organisms when acidified with Cl2, LD50, oral, rat(mg.kg-1): 1100

Zinc chloride - Harmful if swallowed. Causes burning. Highly toxic to aquatic organisms, LD50, oral, rat(mg.kg-1): 350

Formic acid - Corrosive, toxic, flammable

Benzyl alcohol - LD50, oral, rat (mg.kg-1): 1,230

Dichloromethane - Suspected of causing cancer, LD50, oral, rat (mg.kg-1): 1 600 Sulphuric acid, 75 % w/w - corrosive

Dimethylformamide - Flammable liquid and vapor. May harm the fetus in the mother's body. Harmful in contact with skin. Harmful by inhalation. Causes severe eye irritation, Carcinogenicity: May form nitrosamines under certain conditions. Nitrosamines have been shown to be carcinogens in animal tests, LD50, oral: rat = 2200 mg/kg.

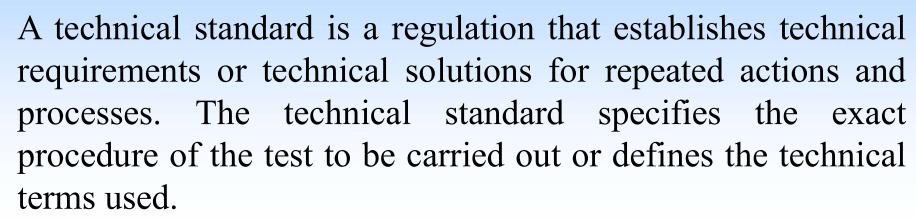
Carbon disulfide - highly flammable liquid and vapour. Skin irritant. Causes severe irritation to the eyes. Suspected of impairing reproductive capacity. Suspected of damaging the fetus in the mother's body. Causes damage to organs with prolonged or repeated exposure. LD50, oral, rat (mg.kg-1): 3 188

Anhydrous (glacial) acetic acid - corrosive

Cyclohexanone - Harmful by ingestion, skin contact and inhalation. Irritating to skin. Causes serious eye damage. LD50, oral, rat (mg.kg-1): 1890

Xylene - Flammable liquid and vapour. May cause death if swallowed and enters the respiratory tract. Irritating to skin. Causes severe eye irritation. May cause respiratory irritation. May cause organ damage by prolonged or repeated exposure. Harmful in contact with skin or by inhalation, LD50, oral,

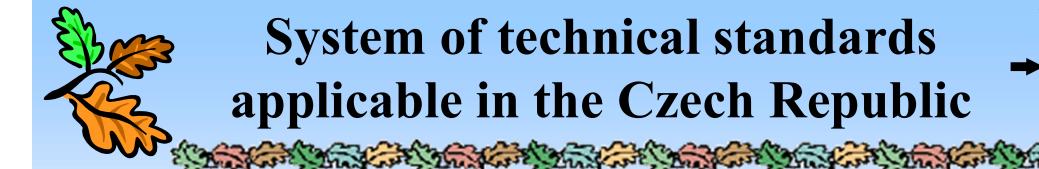
Technical standards



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With a little exaggeration, it can be said that what cannot be measured properly must be standardised. The main objective of standards (or standardisation) is to understand and define. A technical standard is actually a documented agreement.

The use of standardised procedures is in principle voluntary. Some standards have been made binding by law, e.g. some standards defining e.g. occupational safety and health.



- 1. ISO international standards that summarize test methods, markings and terminology
- 2. EN European standards that summarise test methods, marking, terminology and product safety
- 3. CSN Czech (formerly Czechoslovak) standards, national validity. In technical practice, national standards of other countries can also be encountered:

DIN (SRN), GOST (Russia), BS (GB), ASA (USA), ASTM (USA).

Common cloths (textiles)

Textiles fibers are used typically as blends

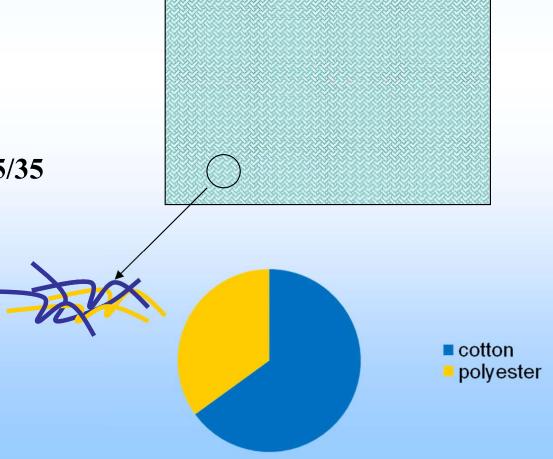
1 textile=more fibers

Typically 2 compounds
Up to 4 -5 different fibers

for example: cotton polyester 65/35

65 % cotton

35 % polyester



Quantitative fiber analyses – problem of sample humidity

The fibers change its weight according the relative humidity of air Ratio of fibers at standard conditions !!!

Standard condition (for business, laboratory...): 65% RH, 20°C

Difference to absolute dry fibers:

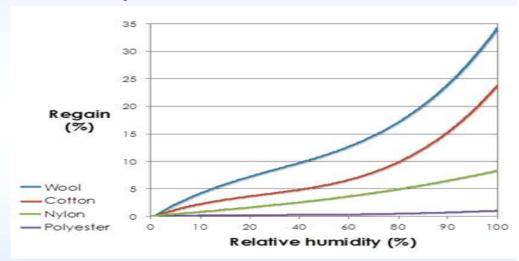
Polypropylene +0.1%

Polyester +0.7%

Polyamide +4%

Cotton +7%

Wool +18%



Natural fibers = problem of humidity

Different humidity - different ratio of components in a blend

Sample: absolute dry polypropylene/ wool 40/60

At standard conditions: ???

40x1/(40x1+60x1.18)=40/111=36%

60x1.18/(40x1+60x1.18)=71/111=64% 36%+64%=100%

Quantitative fiber analyses – problem of sample humidity

Water absorbed in fibers influences the results of quantitative analyses.

Each fiber absorbs different quantity of water according its chemical composition. The quantity of absorbed water is connected with their humidity.

Higher air humidity = high contain of water in fibers

Solving: before the analyses is necessary remove the water from fibers by drying at 105 °C – time is unlimited, we have to wait to stabilized weight, ideal dried (absolute dry) sample has the weight M1

fibers

humidity

Qualitative analyses – fiber identification

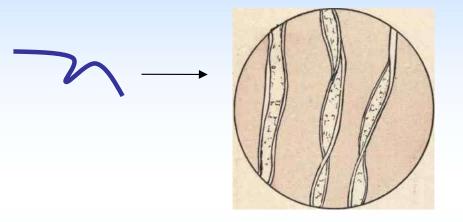
First step = fiber identification

Is it a mixture?

What kind of fibers is inside?

Special fibers (=special problems)

- glass fibers, basalt fibers, aramides...
- nobody has the useful methods
- to stable for standard decomposition methods



Quantitative fiber analyses – problem with impurities

impurities

Possible problems:

- Additives (pigments from prints ...)
- Finishing (hydrophobic treatment, flame retardants ...)
- Preparations
- Solving: Washing out of all impurities (or extraction in organic solvents)
- The behavior of these impurities in solvent is unknown will be soluble completely, partially ... ?
- Impurities (especially final treatments) can reduce the affect of solvent to textile fibers are protected, the wetability is low...
- Preparations we should remove before the analyses typically by organic solvent – application of nonpolar solvent in many cycles (mixture of petrolether and ethylether in ration 1:1)
- Other kinds of impurities is necessary to remove by other method enzymatic desizing, or washing in water

Soxhlet extractor + filtration 10 11 Frita - filtration description pores 150-250 μm **S0** 2 90-160 μm **S1 40-90 μm S2** 15-40 μm **S3 S4** 5-15 μm



Sampling and analysis should be carefully documented, preferably in a detailed laboratory logbook (or analysis book). A proper record should include answers to the questions Who?, Where?, What?, When?, How many times?, According to what?, How?, What was the result?

A carefully kept laboratory logbook is useful for tracing back information, e.g. on variations in the properties of raw materials, products, etc.

In laboratories, input tests of raw materials, intermediate control and output tests of products are often carried out. This information allows to ensure stable production quality and to identify sources of problems.

The transfer of raw materials, semi-finished products and products in industry is based on agreements between suppliers and customers, who define by mutual agreement the characteristics of the material to be transferred. In general, the supplier tries to have loose standards and the buyer tries to tighten the standards.



Samples for analysis should be taken so that they are averaged and defined (!).

For example, when taking wool fibres from a bale, samples should be taken not only from the surface of the bale but also from deeper parts of the bale.

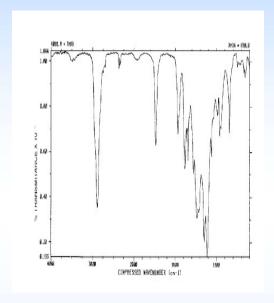
When checking the yardage, the beginning of the middle and the end of the lot should be observed. In general, the edge and centre of the fabric (in the direction of the weft) may also vary.

For textile processing aids (TPPs), it is often necessary to homogenize the chemical throughout the container. This is a very complicated operation when removing, for example, fatty substances. For reasons of a deeper understanding of the processes involved, e.g. separate layers in the container can also be tested.

Quantitative fiber analyses – possibility methods

Quantitative analyses:

- <u>Instrumental methods such as IR spectroscopy</u> special applications, short time results, low accuracy



- <u>Destructive weight analyze methods</u> standard technique, robust, long time experiments



Quantitative fiber analyses

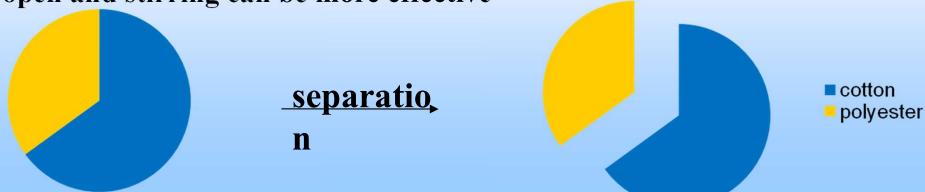
Destructive weight analyze methods

<u>Basic idea:</u> I remove only one component from the fiber blend, fiber rest will contain other kinds of fibers. The weight of these fibers is the same as in original sample.

The weight of original sample we know – we can calculate the contain of the removed fibers in the original sample.

So we can continue with all other fibers. The sum of all fibers in one textile is 100%.

For analyses is recommended to cut sample to small peaces (approximately 10x10mm) Smaller samples are more sensitive to used solvents – the structure is more open and stirring can be more effective

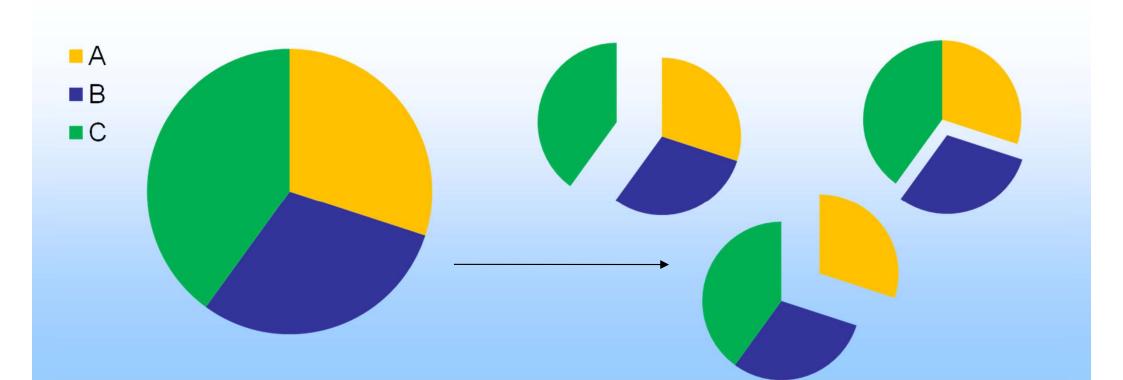


Quantitative fiber analyses

All times we are using methods, with remove only one kind of fibers from the textile.

Mixture of 3 (or more) components:

Is necessary to use 2 (or more) methods, one method for dissolving of one fiber For every test is necessary to use new sample. Using of before used (attacked) fibers is prohibited – used fibers are more sensitive to new solvent.



Quantitative fiber analyses - organization

Standard recipe:

- 1) qualitative analyses of fibers (aim: is it a blend? Find the right method of qualitative analyses)
- 2) preparation of tree parallel samples, 2g each
- 3) cleaning of samples extraction
- 4) cutting of samples to small peaces
- 5) drying (105°C) of samples to constant weight M1
- 6) dissolving of "more soluble" component of blend according the technical standard
- 7) the rest of insoluble fibers separate quantitatively from the solution on the filter glass
- 8) wash out the insoluble of solvent from the residual fibers
- 9) drying of insoluble fibers up to constant weight M2
- 10) calculation of the ratio of both components in this blend in the case of absolute dry samples
- 11) calculation of the ratio of both components in this blend in the case of standard air-conditioned samples

Methods for quantitative fiber analyses

Fiber blend		method	
soluble	insoluble	chemicals	condition
cotton, polyamide, viscose	Polyester, acrylic, polypropylene, wool	H2SO4 75%	50°C, 1 hour
acetate	Other fibers	acetone	20°C, 1 hour
acrylic, polyvinylchloride, acetate	Other fibers	DMF	95°C, 1 hour
polyamide, acetate	Other fibers	HCOOH 85%	20°C, 15 min
wool, silk	Other fibers	NaClO (35g active Cl/litre) + 5g/l NaOH	20°C, 40 min
polyvinylchloride	Other fibers	cyklohexanon	Boiling temperature, 60 min

Calculation of fiber ratio (dry)

calculation of the ratio of both components in this blend in the case of absolute dry samples:

Ratio of dissolved fibers (a1) in % in absolute dry sample:

$$a_1 = \frac{(m_1 - m_2.K).100}{m_1}$$

Dry! m1

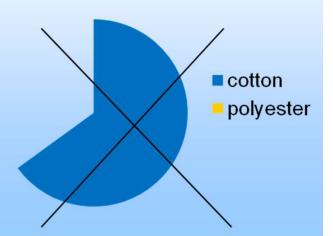


Ratio of insoluble fibers (a_2) in % in absolute dry sample:

$$a_2 = \frac{(m_2.K).100}{m_1}$$

m₁ weight of all fibers together in g,
 m₂ weight of insoluble fibers v g,





Correction factor K

K ... correction factor, "insoluble" fibers are sometimes "low soluble" in the used solvent, factor is in technical standard

The methods are selected so, that the factor is typically 1.

Ideal K=1:

One kind of fibers is total dissolved

The other component is not attacked

Ratio of air-conditioned fibers

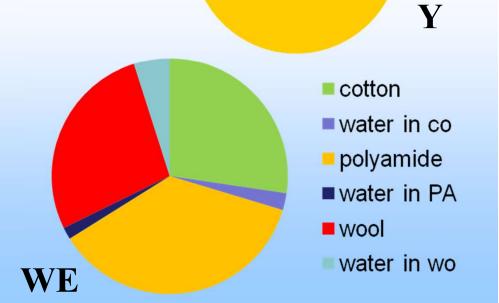
calculation of the ratio of both components in this blend in the case of standard air-conditioned samples (65% air relative humidity, 20°C):

$$X_1 = \frac{a_1.HF_1}{a_1.HF_1 + a_2.HF_2 + ... + a_n.HF_n}$$
 $X_2 = \frac{a_2.HF_2}{a_1.HF_1 + a_2.FV_2 + ... + a_n.HF_n}$

 X_1 ... weight ratio of fibers 1 in analyzed textile at standard conditions

 X_2 ... weight ratio of fibers 2 in analyzed textile at standard conditions

fiber	Humidity of fiber (HF)	
Wool	1.180	
Viscose	1.110	
Polyamide	1.045	
Cotton	1.085	
Acetate	1.060	
polyvinylchloride	1.010	
acrylics	1.010	
polyester	1.007	
polypropylene	1.001	



cotton

■ wool

polyamide

DR

Real sample

Real sample:

Mixture cotton/polyester (estimated by microscopy and qualitative fiber analyses)

Weigh of air dry sample is 2.458 g (high precision is necessary)

After drying at 105°C we have: M1=2.327 g

The humidity in textile was: H=(2.458-2.327)/(2.458)=5,32%

(it is a redundant calculation – we don't know the humidity of air and the real property of fibers, damaged fibers or fibers after chemical modification can have different standard humidity)

Samples we will cat to small peaces – to intensify the solution

Real sample

Real sample:

In the mixture cotton/polyester is more sensitive to solvents cotton – we will use the solvent for cotton 75% H2SO4

We will heat the 75% H2SO4 up the 50°C in a thermostatic bath

We will put in our cuted sample

We will wait for 1 hour, during this time we will mix this reaction mixture well

The separation of insoluble fibers on filter glass (=sintered glass filter, =sintered glass), the absolute dry weight of filter glass is 15.456g

Rinsing of insoluble fibers on the filter glass with solution of

Real sample

Real sample:

Calculation of insoluble fibers weight (absolute dry) M2= 1.333g

Calculation of ratio between absolute dry components:

Cotton: $a1 = (2.327 - 1.333 \times 1)/2,327 = 42.7\%$

Polyester: a2 = (1.333x1)/2,327 = 57.3%

a1+a2=100%

Prediction of ration of cotton and polyester as standard airconditioned samples (65% air relative humidity, 20°C):

Cotton: $a1 = (2.327 - 1.333 \times 1.007)/2,327 = \%$

